

Zalāa-2. 'vāv šy ātatāā šyl āvāo āvāh' nī

Write down method of manufacture of chloral.

OR

Nāvakāšyē/ā āšyūā; āšyl 'āmšyl 'wpašyūāvāo ēāāē/ā yālm ytl āčy nī

Explain with example kinetics and mechanism of halogenation reaction.

Zalāa-3. rgj 'wpyym ylyāyāšyē/ā āšyūā; āšyl āvāāxmā ytl āčy nī

Explain characteristics of batch and continuous sulphonation.

OR

Āānvāā šy ylyāyāšyē/ā šyl āšyūāvāo ytl āčy nī

Explain the mechanism of sulphonation of Naphthalene.

Zalāa-4. wāwāū ēqj āē šy ēqšyē/āwšyā w/āā šylāk' nī

Describe equipments involved in Aerobic treatment process.

OR

i āōlāā'ā 'wpaēāšy; āāūāā šyāytl āčy nī

Explain adsorption and its applications.

Zalāa-5. 'py; qālā pZar'āā šyl āvāo šyā w/āā šylāk' nī

Describe the method of solid waste management.

OR

ātD'p'vātāāē (ōb āāšyāšy) šy āy÷ān 'wpašyūāvāo šyl āwvj āā šylāk' nī

Discuss the principle and working of mist eliminators.

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Annual Examination - 2017

B.Sc.-II

INDUSTRIAL CHEMISTRY

Paper - II

Max.Marks : 33

Min Marks : 11

Time : 3 Hrs.

'iq β h'p'i ; tθ8 ; ānvi ēēā Zalāa Nā āk'āw'āšyē/āā ; āāwāūēā h'p'r' tēvi ēēā Zalāa h'p'y' tēāāi ēēēāū Zalāa Nā h'p'i ; 'šyāyryçqñvçñv šyēñ

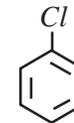
Note: Section 'A' , containing 8 very short answer type questions, is compulsory. Section 'B' consists of short answer type questions and Section 'C' consists of long answer type questions. Section 'A' has to be solved first.



āāāāšyē/ā ; ān vi ēēēāū Zalāa šy ēēēā 'šyā' uā āāwā'ūāp tēāāēñ (Answer the following very short-answer-type questions in one or two sentences) (1x8=8)

Zalāa-1. āāāā ; āāšyūā qāēšylāk' β

Complete the following reaction :



Zalāa-2. āāē t'p'āā ; āāšyūā šy çvç' t'p'āçv ' 'uā Nē?

What is the electrophile of nitration reaction?

Zalāa-3. āāāā ; āāšyūā qāēšylāk' β

Complete the following reaction :

